

Characterization of an innovative method for RuO₂ deposition using Electron Microscopy

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Abstract

Many photocatalysts work better or exclusively when a suitable cocatalyst, such as RuO₂, is deposited on their surface. An innovative method of RuO₂ deposition has been found to improve the performance of photocatalysts such as (Ga_{1-x}Zn_x)(N_{1-x}O_x), WO₃, SrTiO₃ and TiO₂. Here we use high angle annular dark field (HAADF) imaging, energy-dispersive X-ray (EDX) spectroscopy, and electron energy loss spectroscopy (EELS) in the scanning transmission electron microscope (STEM) to study the deposition of RuO₂ on TiO₂. The deposition process occurs in two steps, for each of which we are able to characterize the RuO₂ distribution, morphology and crystallinity.

Deposition method

- Deposition was carried out onto metal oxide powders, silicon wafers and glass substrates.
- Each substrate was coated with an organic linker molecule.
- The functionalized sample was then exposed to RuO₂, which binds to the linker.
- Calcination at 350°C was used to burn off the linker, leaving only RuO₂ on the substrate.
- Photolithography was used to pattern the substrate to define regions where deposition should occur.

Results

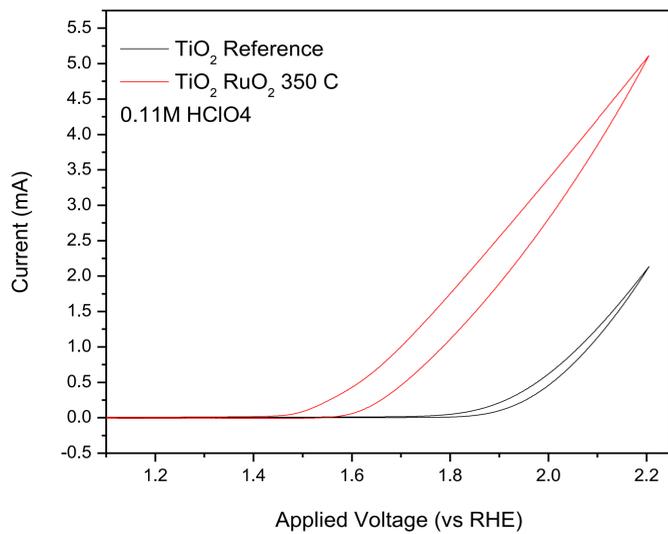


Figure 1. Oxygen Evolution Reaction (OER) performance plotted as a function of applied voltage with respect to the reversible hydrogen electrode (RHE). The red curve shows the improved activity of the coated and calcined sample as compared to bare TiO₂.

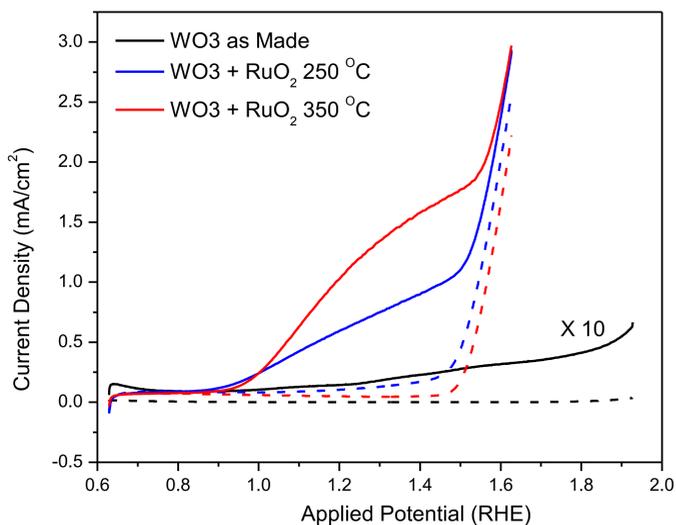


Figure 2. Cyclic Voltamogram acquired from electrodeposited WO₃ on FTO with RuO₂ calcined at 250°C (blue) and 350°C (red) and without RuO₂ (black) in 1M HClO₄. Dashed lines show the response in the dark, while continuous lines show the photoresponse. The photocurrent from the as made WO₃ (black) is scaled to 10 times its original value.

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Characterization

- STEM HAADF imaging is sensitive to variations in atomic number (Z-contrast). Therefore, it allows direct visualization of the RuO₂ deposits.
- STEM EDX and EELS maps and spectra can be used to acquire compositional information from lines or two dimensional regions of the sample.

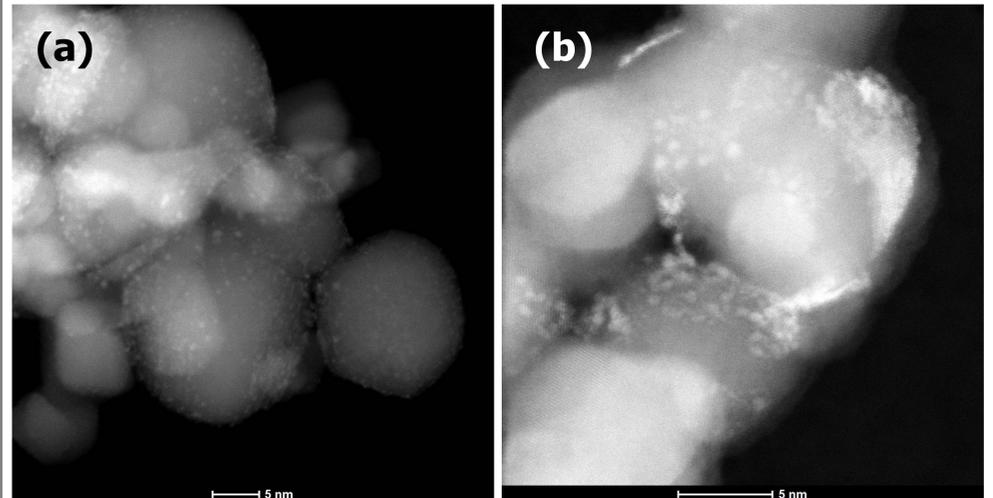


Figure 3. STEM HAADF images of (a) uncalcined and (b) calcined samples. RuO₂ particles appear brighter, while the darker large grey areas are TiO₂. In (a) the particles cover the substrate homogeneously, whereas in (b) some segregation is noticeable. The images suggest that migration across the surface may occur during the thermal treatment.

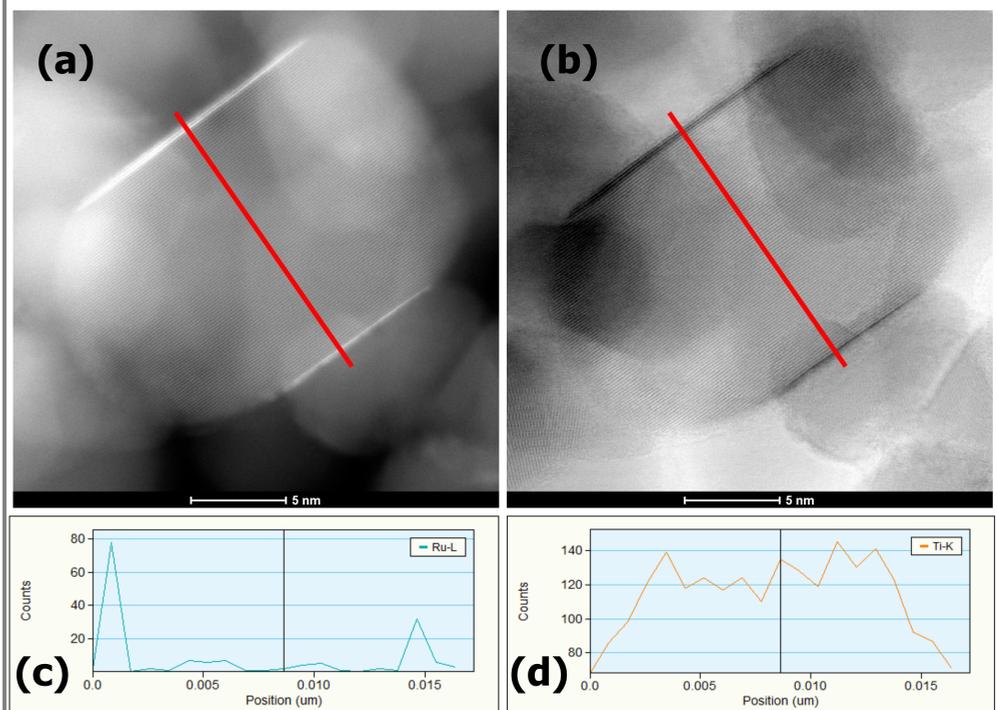


Figure 4. (a) STEM HAADF and (b) corresponding bright field image of a RuO₂ coated TiO₂ particle. (c-d) Ruthenium and Titanium EDX spectrum profiles acquired along the red line.

Conclusions

- The RuO₂ deposition method is effective and yields a uniform distribution of RuO₂ nanoparticles on the substrate before calcination. Their average size is 0.8 nm.
- After calcination the RuO₂ particles sinter into larger clusters (1-3 nm) and form uniform platelets that cover large areas of the substrate (10-200 nm² in area and 0.5 to 1 nm in thickness).
- The segregation process appears to be relevant for the catalytic properties.
- Future TEM investigation will include the *in situ* study of the calcination process during heating and exposure to a controlled gas atmosphere.